

Sensing Superfund Chemicals with Recombinant Systems

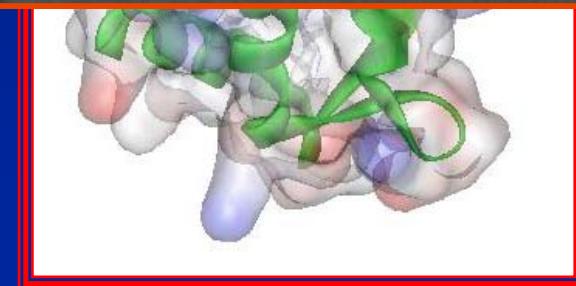
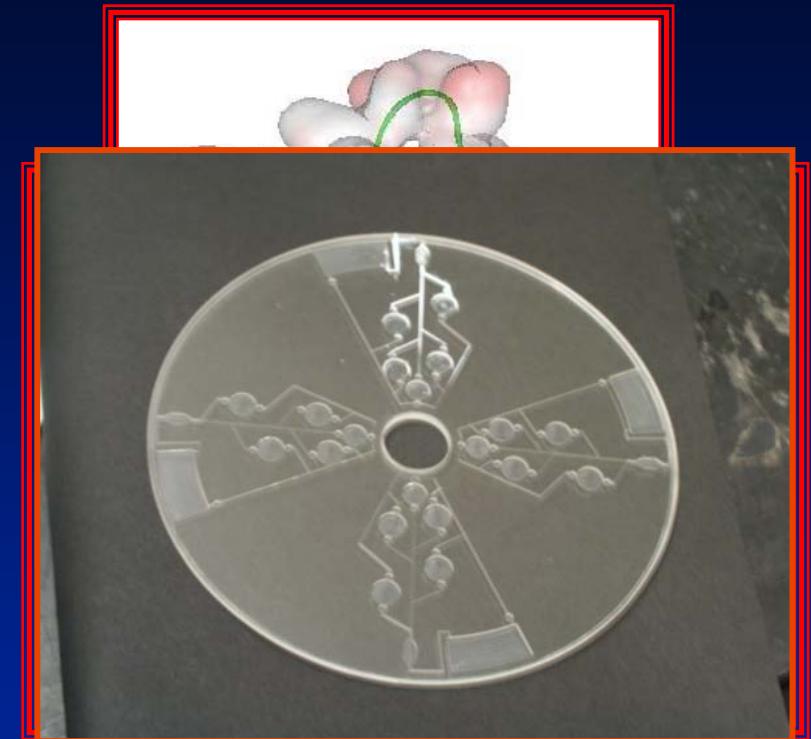
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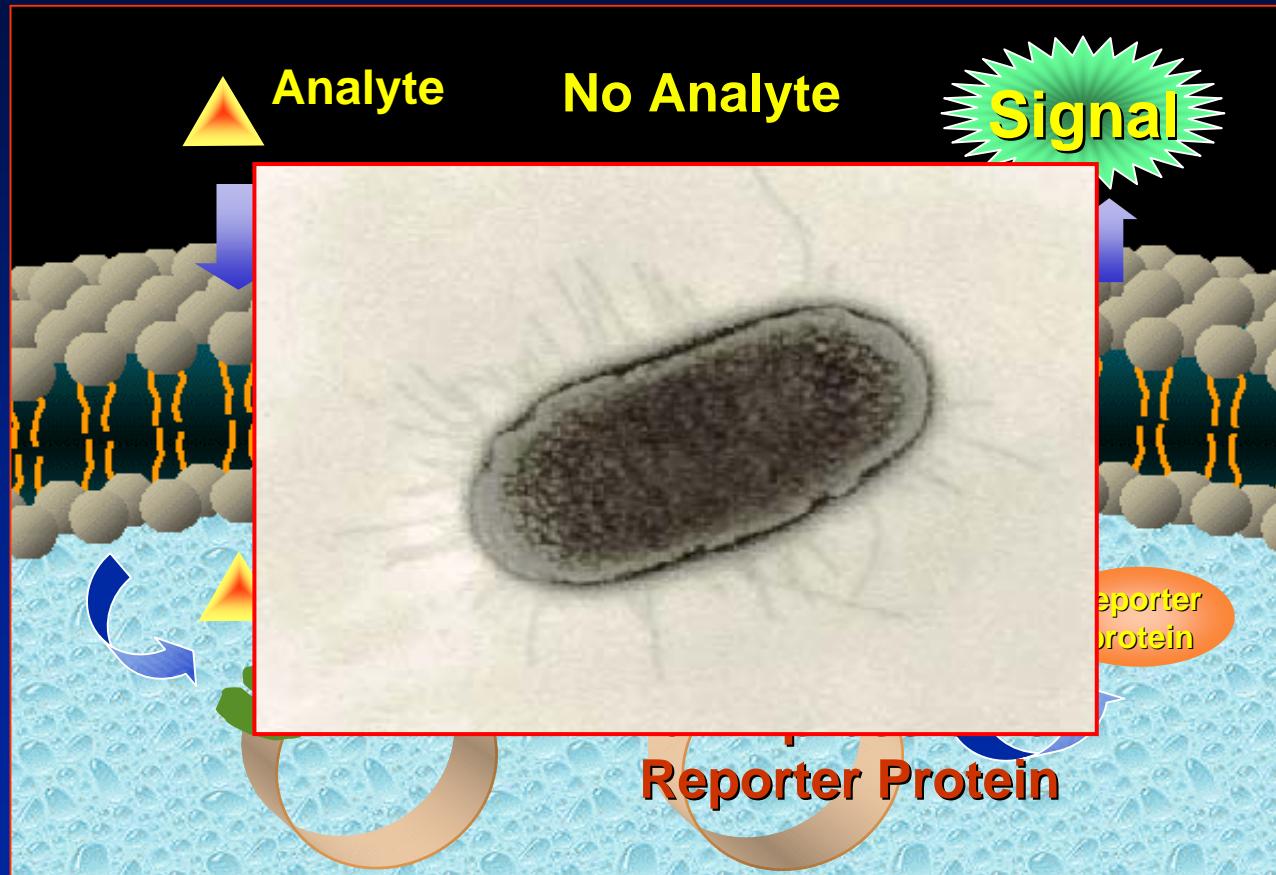


Molecular Recognition in Analytical Chemistry

- Proteins
- Cells
- High Throughput Screening



•Whole Cell-Based Sensing Systems



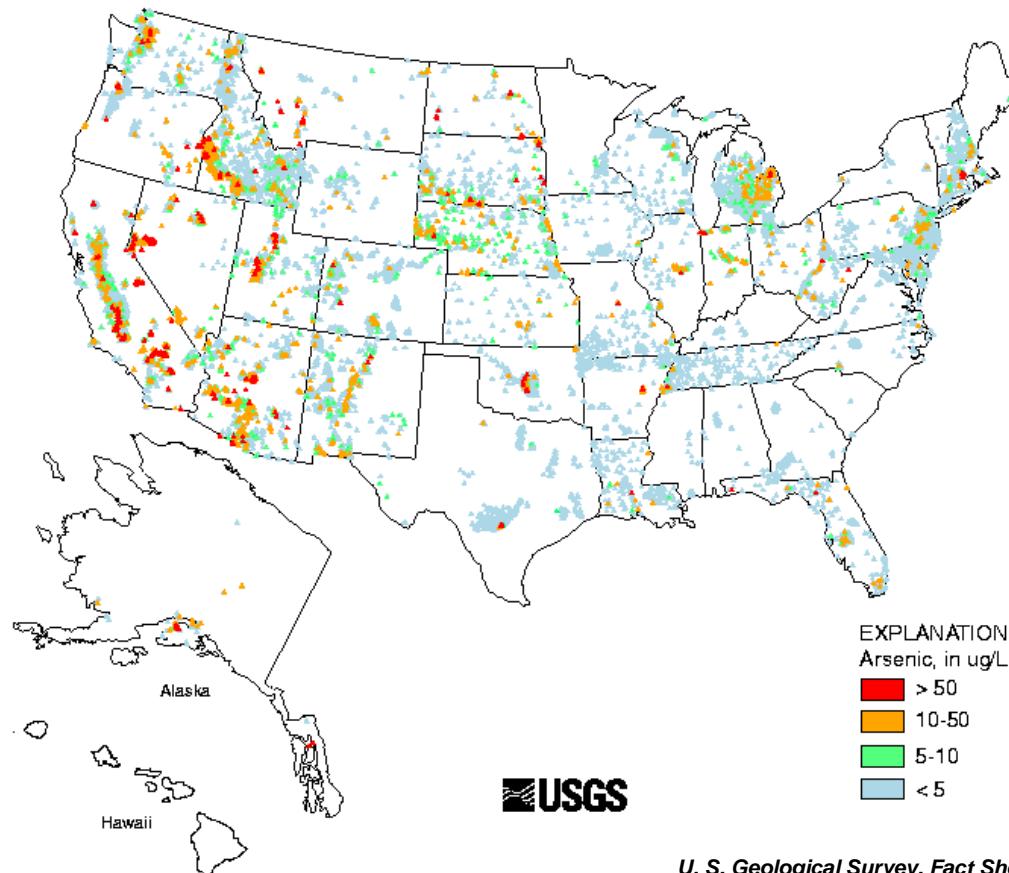
Arsenic Poisoning

- Applications
 - Agriculture
 - Treatment for diseases
 - Industrial uses
- Long exposure to low doses of arsenic
 - Skin hyperpigmentation and cancer
 - Other cancers
 - Inhibition of cellular enzymes



*New Bangladesh Disaster: Wells that Pump Poison... New York Times
November 10, 1998*

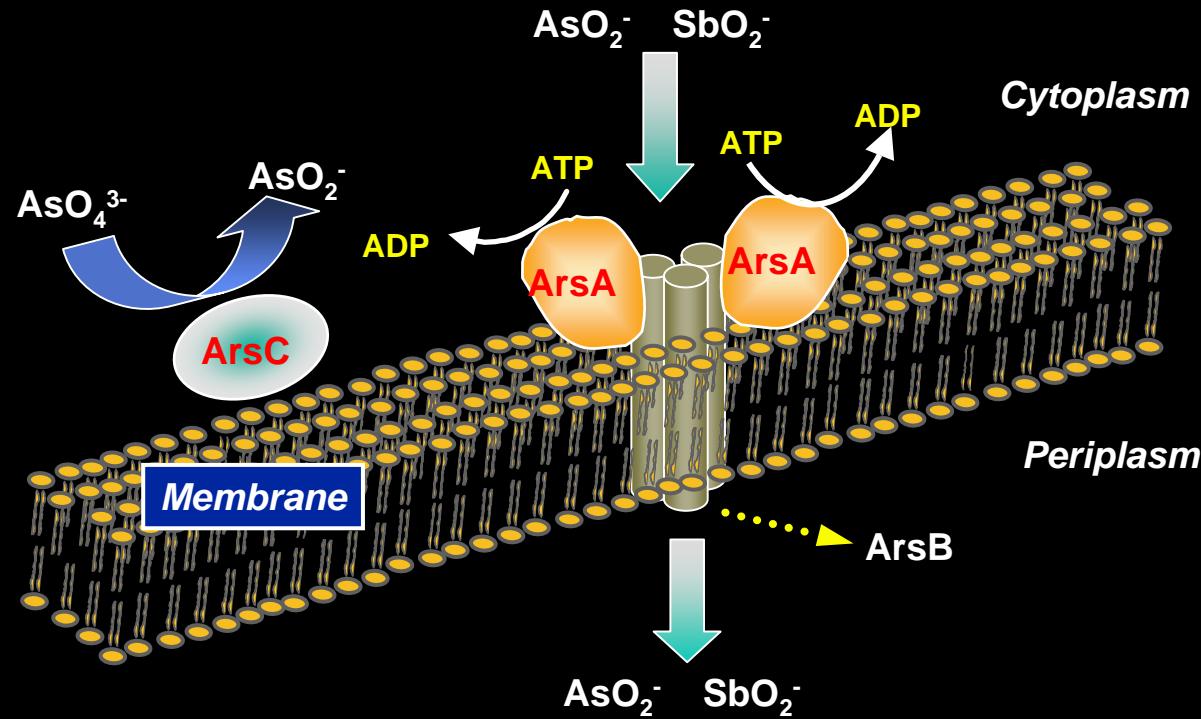
Arsenic contamination in the USA



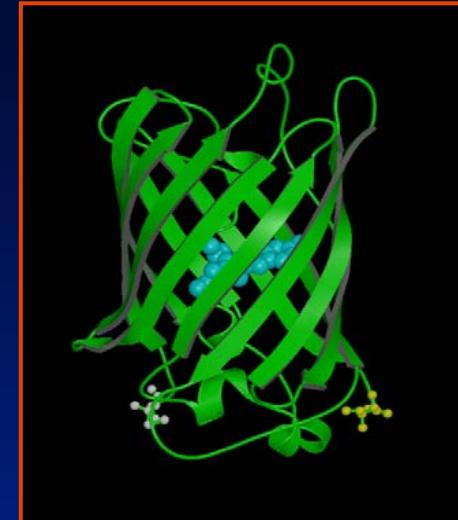
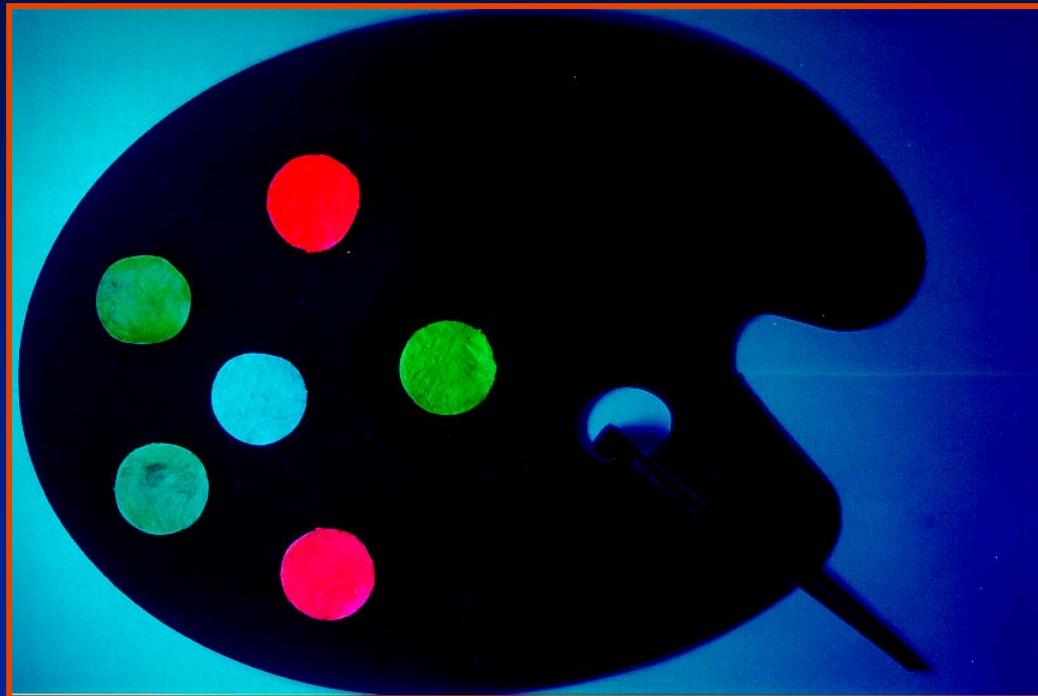
Arsenite Resistance in *E. coli*



Schematic Representation of the Antimonite/Arsenite Pump

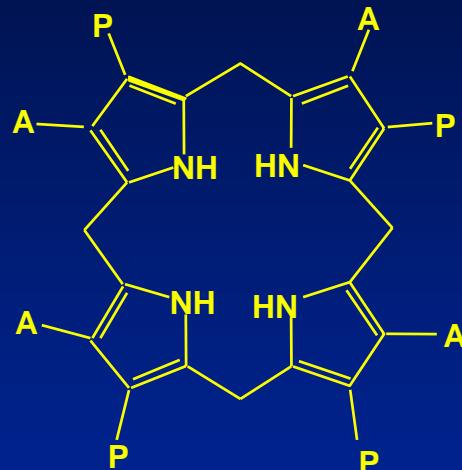


Fluorescent Reporter Proteins in Array Detection

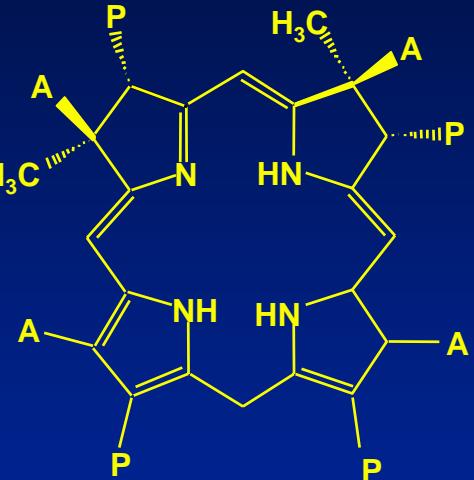


Protein	Excitation λ_{max}	Emission λ_{max}
GFP	395 (470)	509
EGFP	488	509
BFP	380	440
GFPuv	395	509
YFP	513	527
CFP	433	475
CobA	357	605
RFP	558	583

Production of fluorescent porphyrinoid compounds

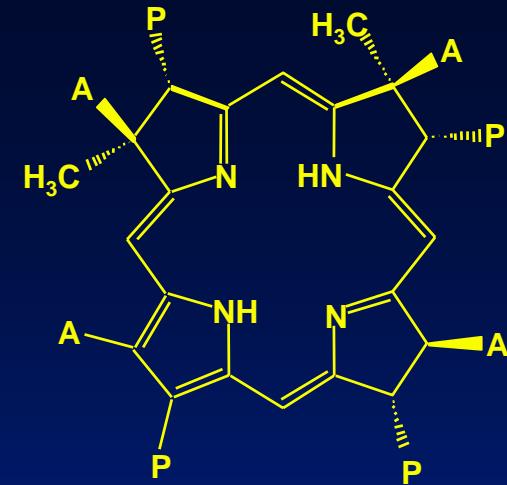


UMT
SAM



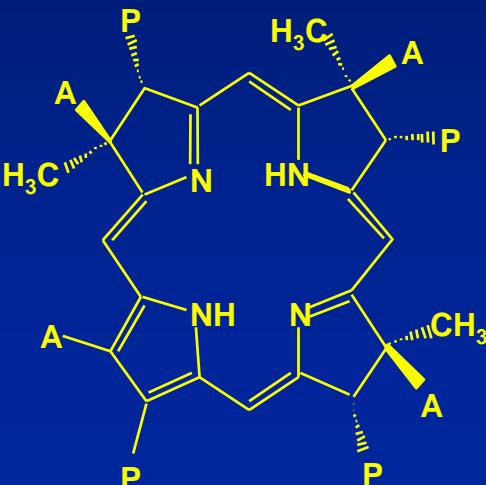
Dihydrosirohydrochlorin
(Precorrin-2)

oxidation



sirohydrochlorin

UMT
SAM



trimethylpyrrocorphin

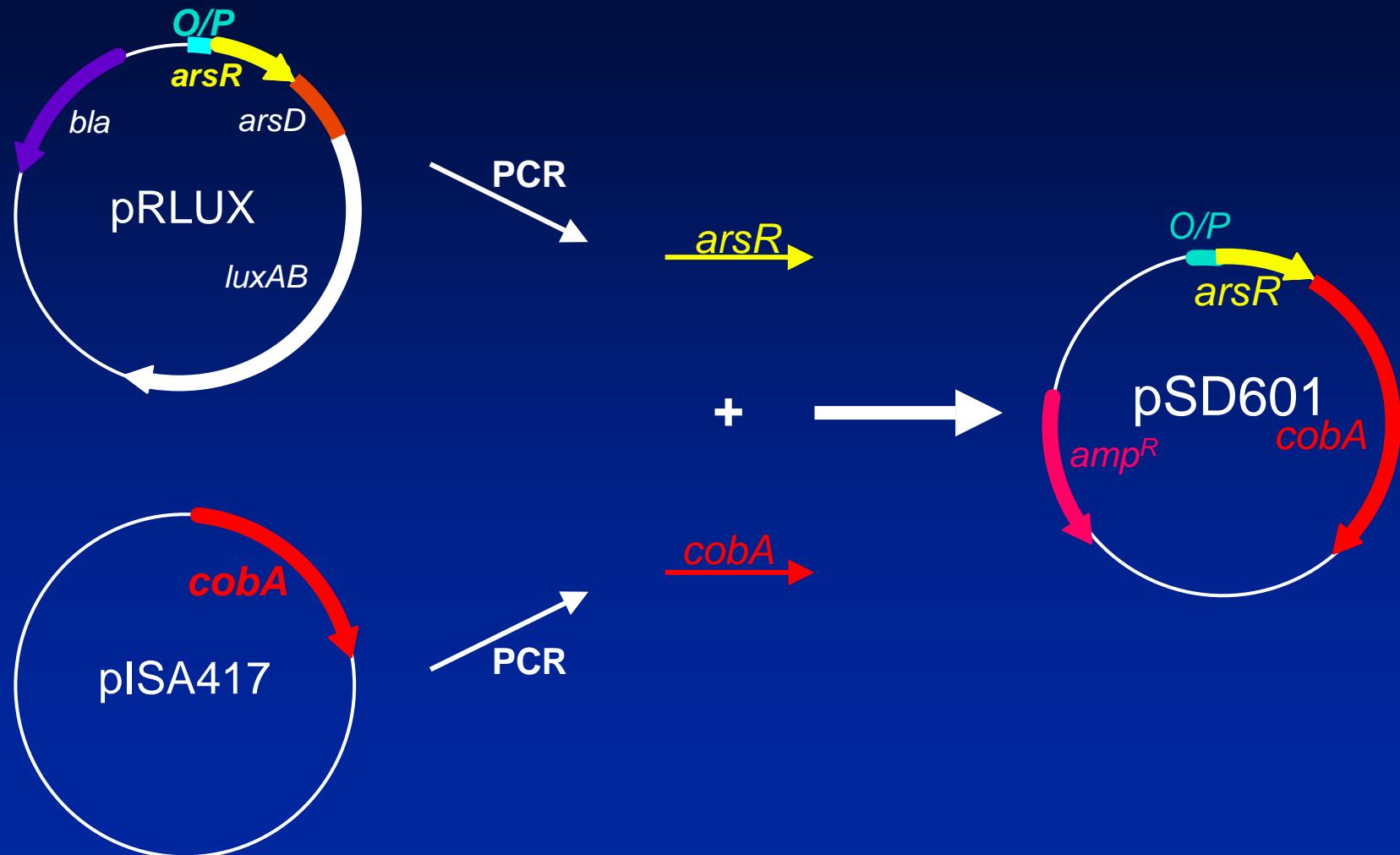
A-CH₂COOH

P-CH₂CH₂COOH

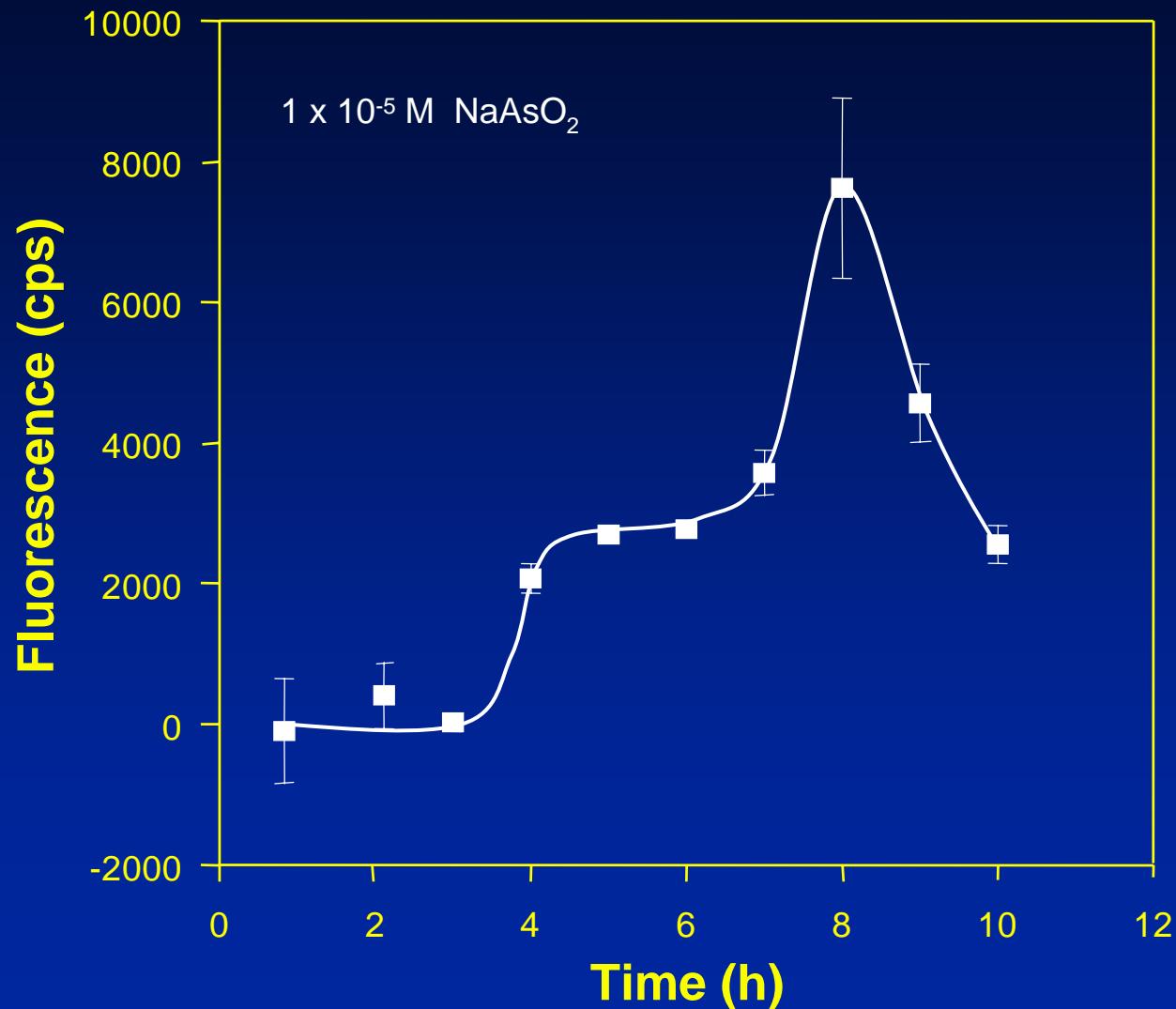
SAM - S-adenosyl-L-methionine

UMT- uroporphyrinogen methyltransferase III

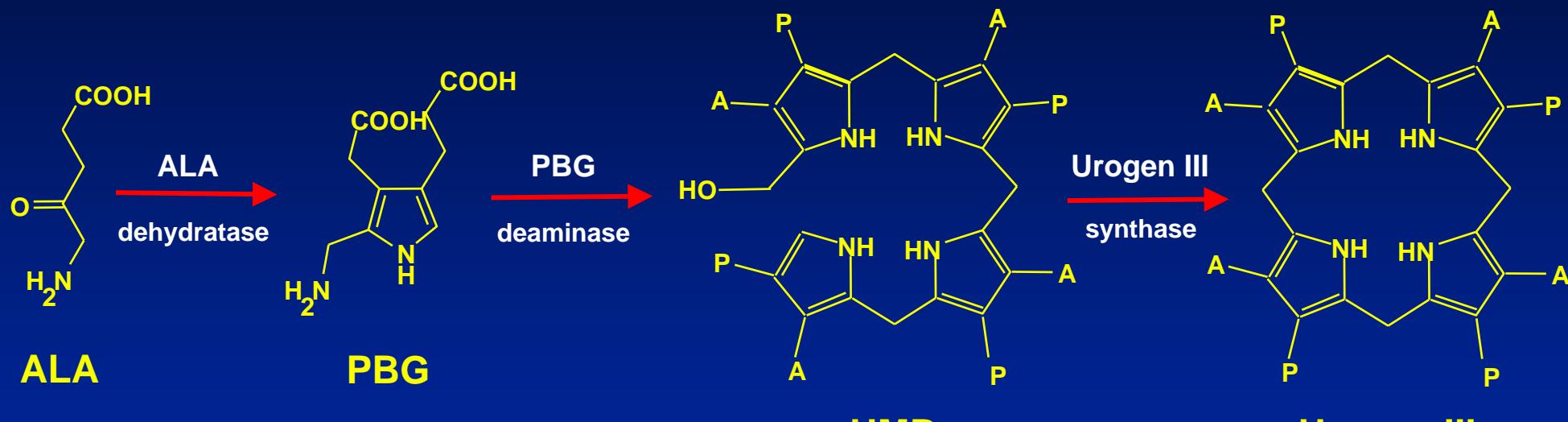
Construction of pSD601 plasmid



Time Study for Arsenite



ALA in Urogen Pathway



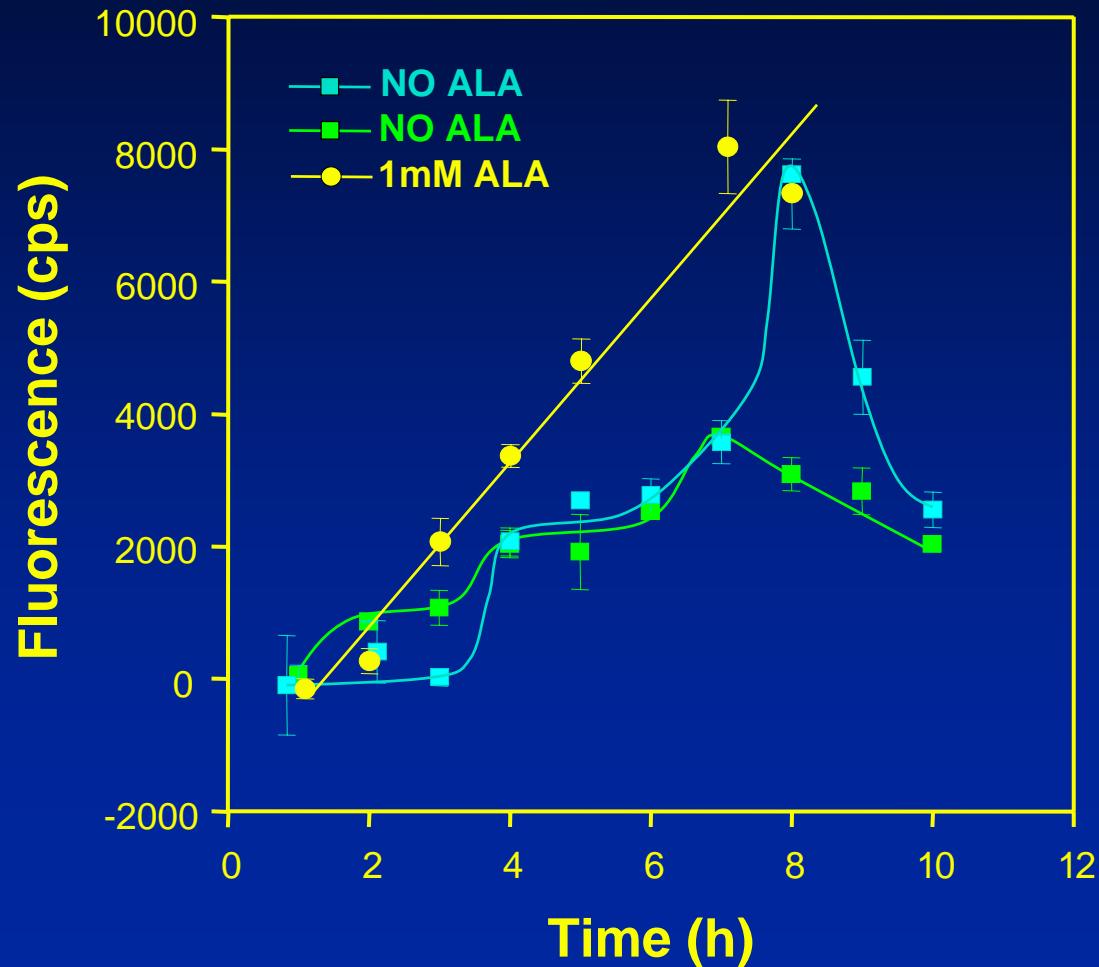
ALA - δ-aminolevulinic acid

PBG - Porphobilinogen

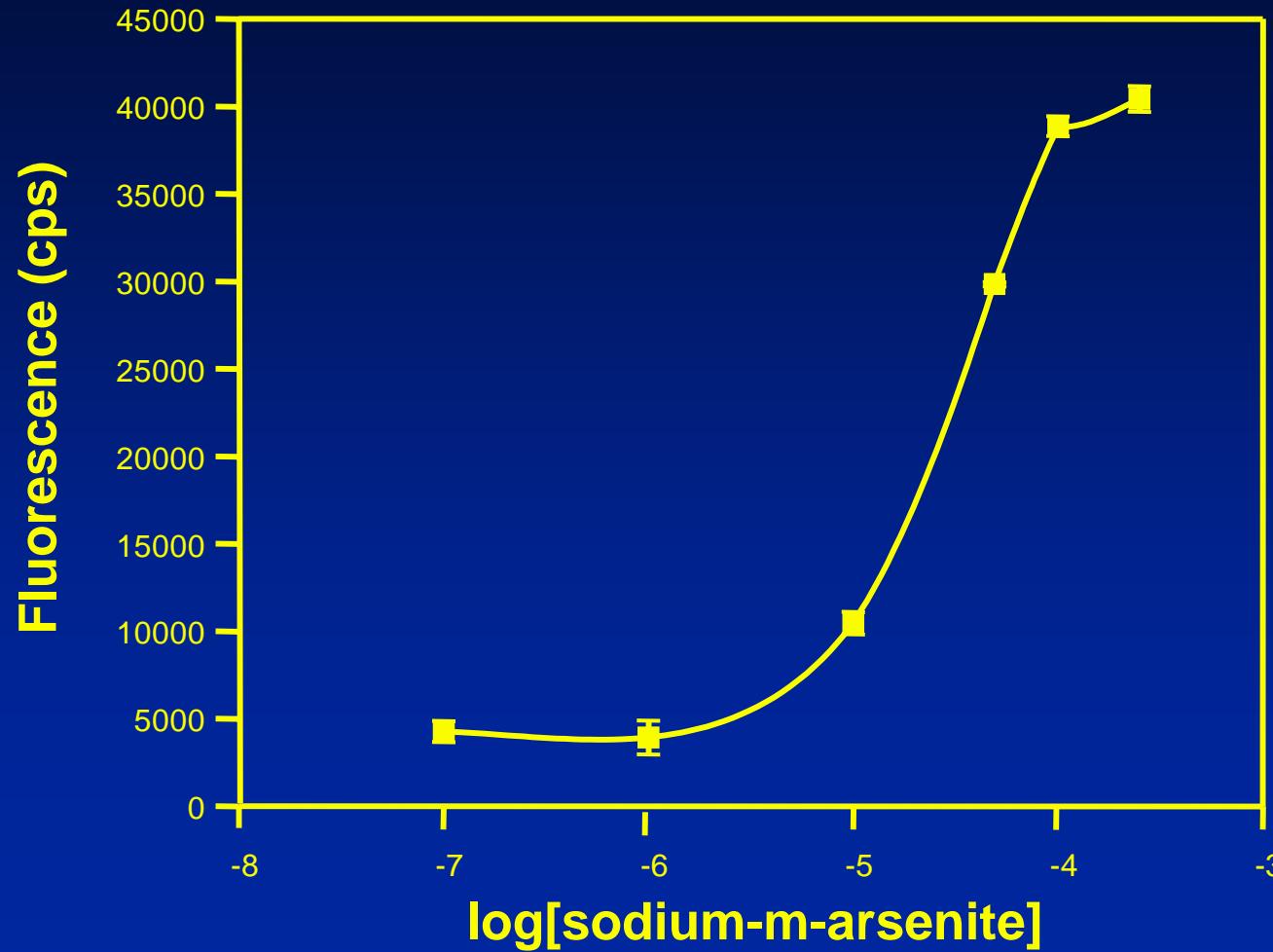
HMB- Hydroxymethylbilane

Urogen- Uroporphyrinogen

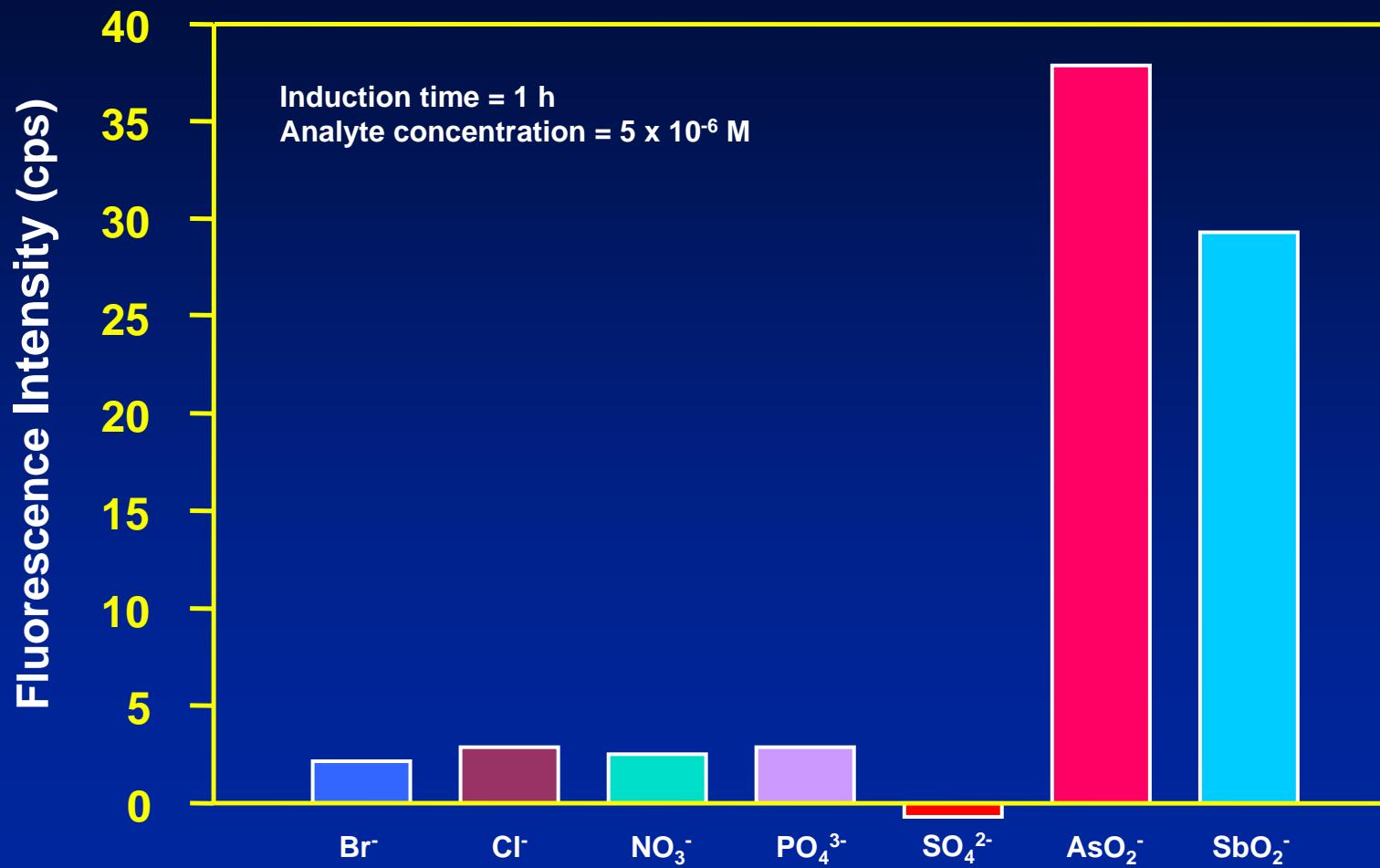
Time Study of Arsenite with ALA



Calibration Plot with ALA



Selectivity Study



Field Challenges

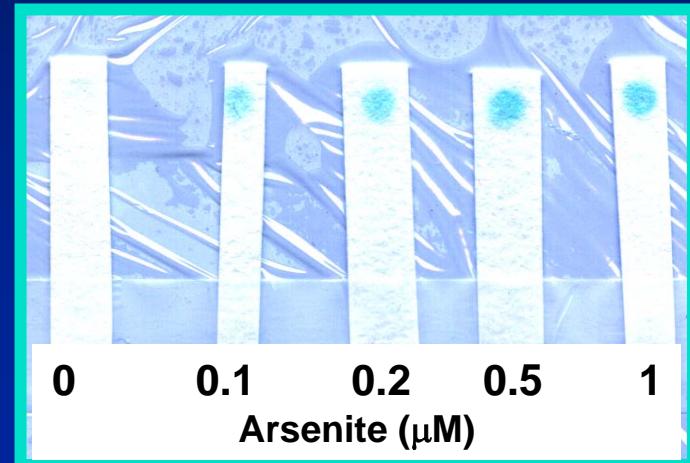
- Conventional analytical techniques
- Field-kits
- Background signal
- Viability of the cells
 - Freeze drying
 - Strips (β -galactosidase)

Addressing Environmental Analysis with Self-Contained Kits



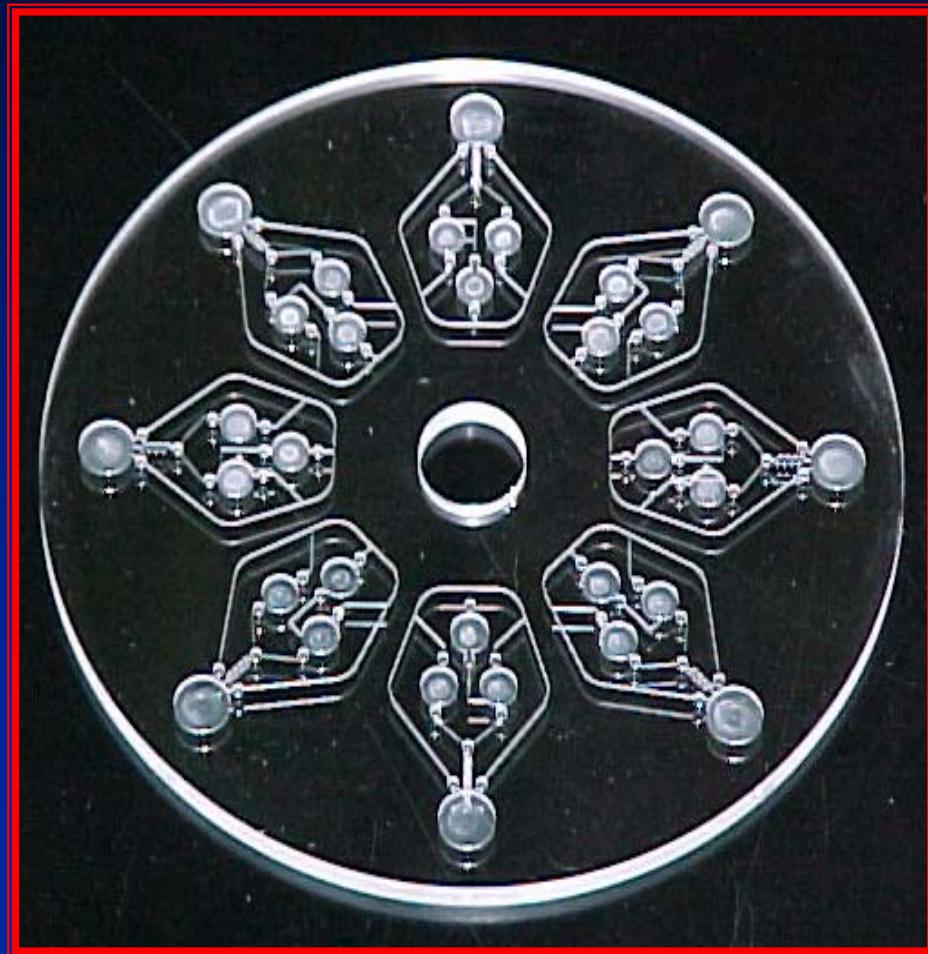
<http://www.nature.com/nsu/030929/030929-7.html>

Lyophilized
Reagents



Strip Sensors

Centrifugal Microfluidic Platform for Micro-Total Analysis Systems

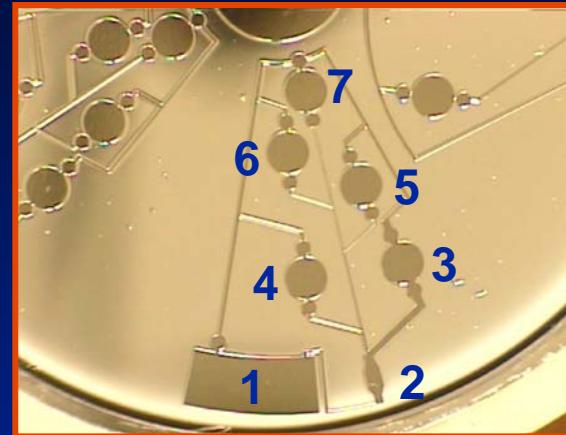


- Low power and space requirements**
- Less reagent and sample consumption**
- Easy disposal**
- Short analysis time**
- Integrate washing, sample preparation, and calibration**

Prototype Compact Disc

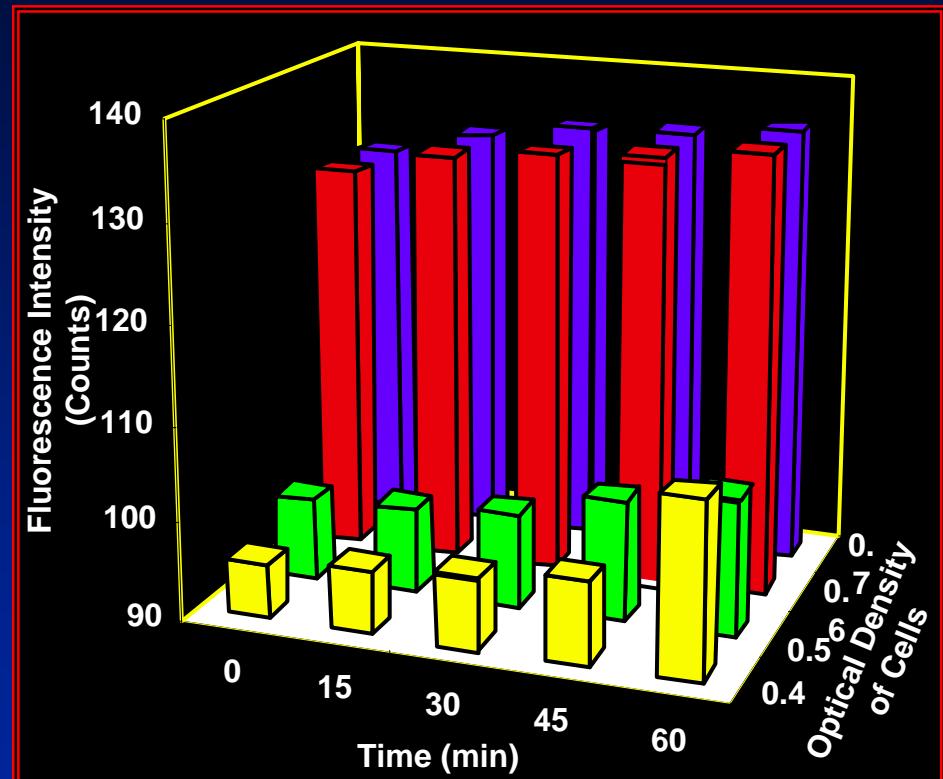
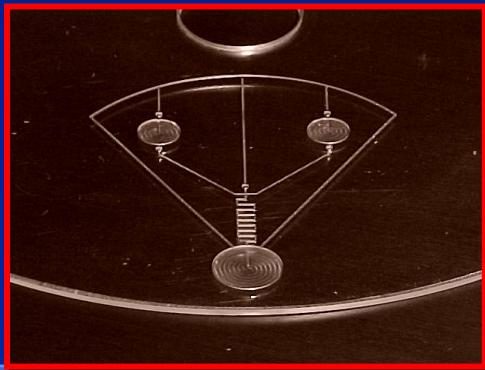
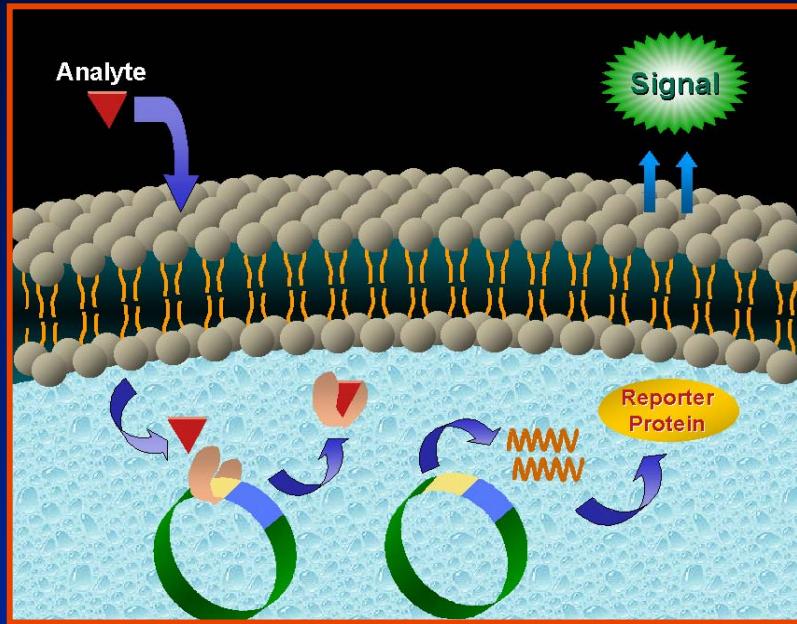


Prototype CD for four simultaneous analyses

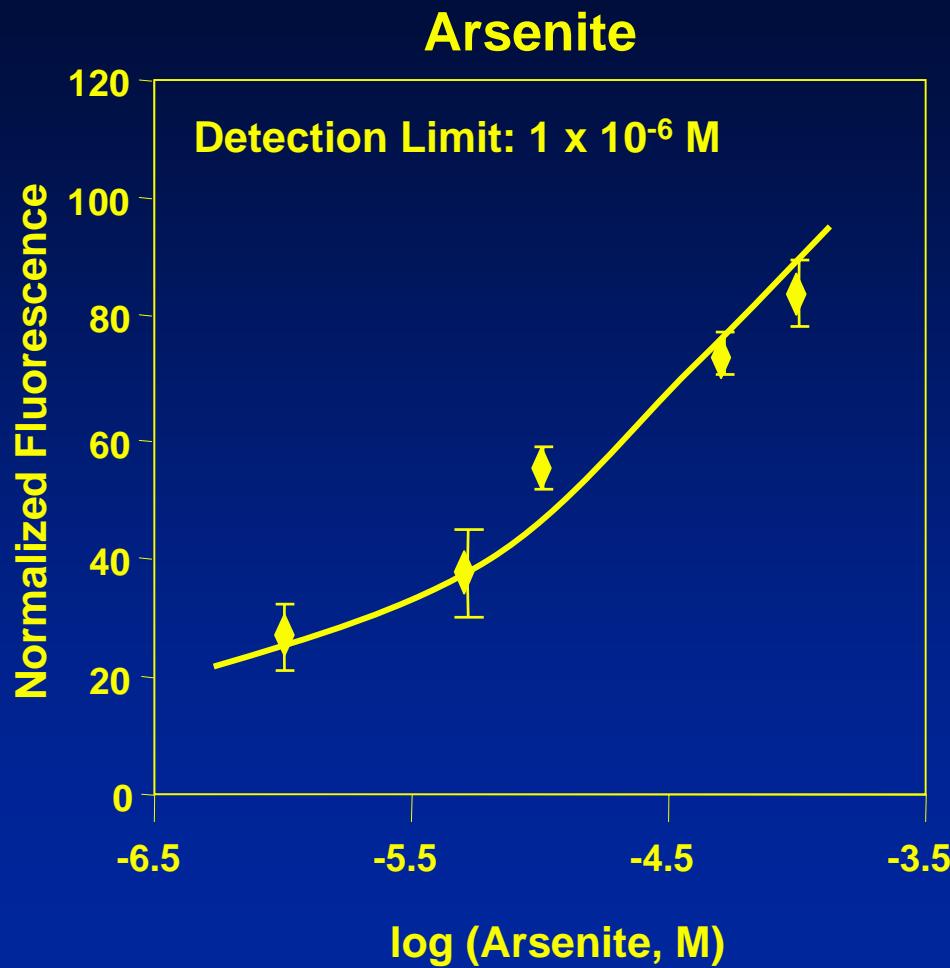
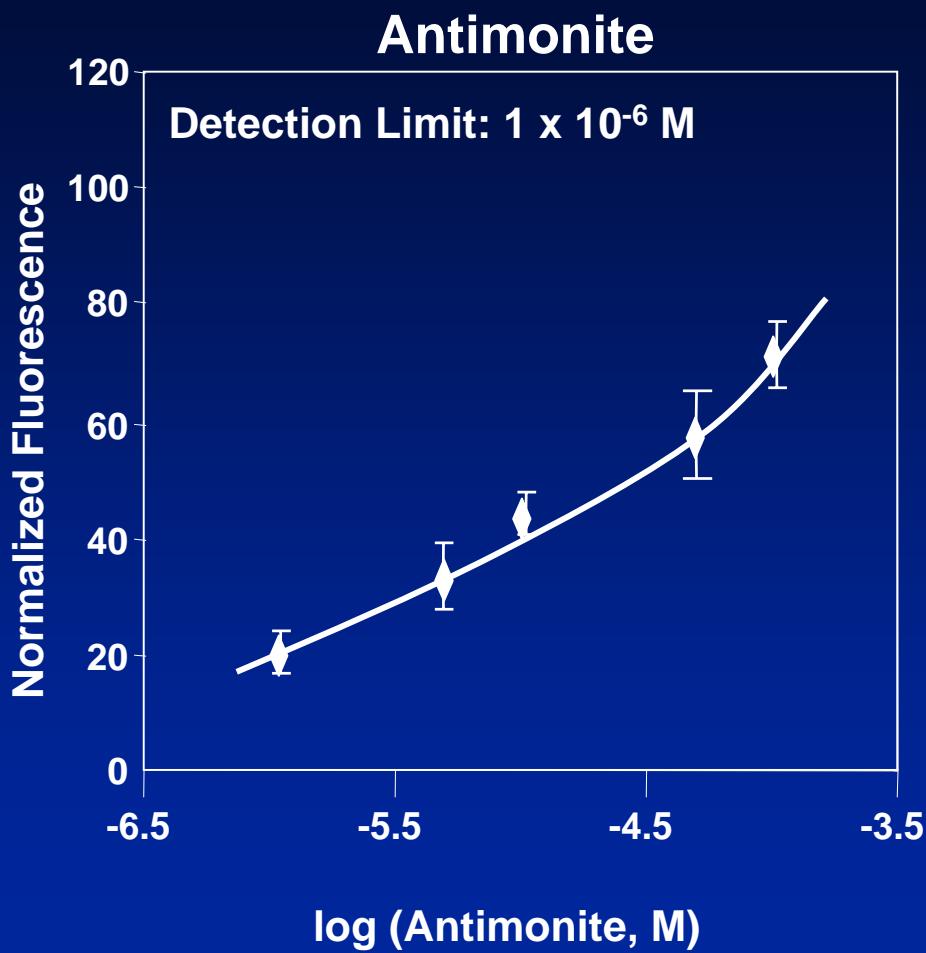


1. Waste Reservoir
2. Optode
- 3-7. Solution Reservoirs

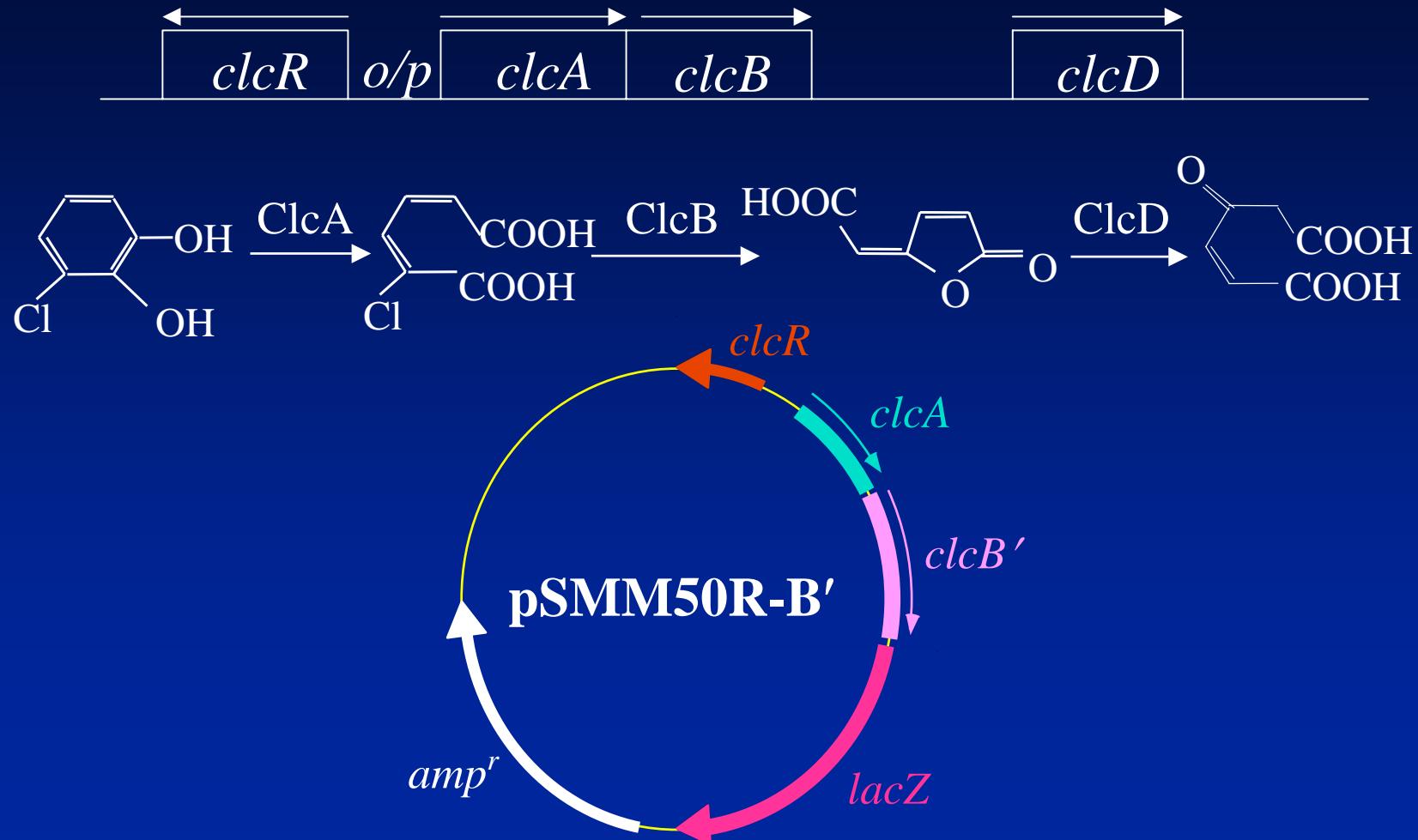
Incorporation of Whole-cell Sensing System for Arsenite/Antimonite on the CD Platform



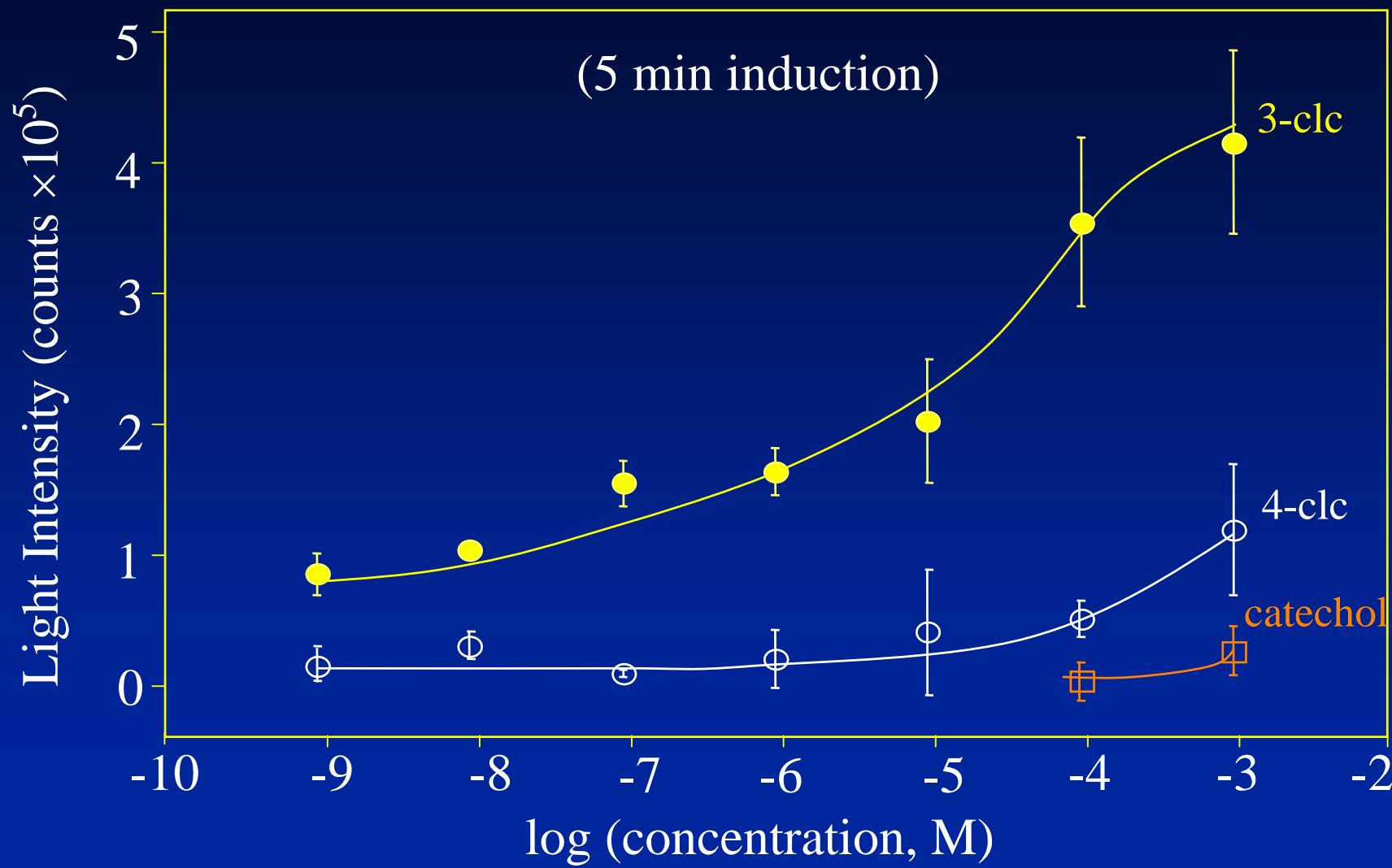
Calibration Curves



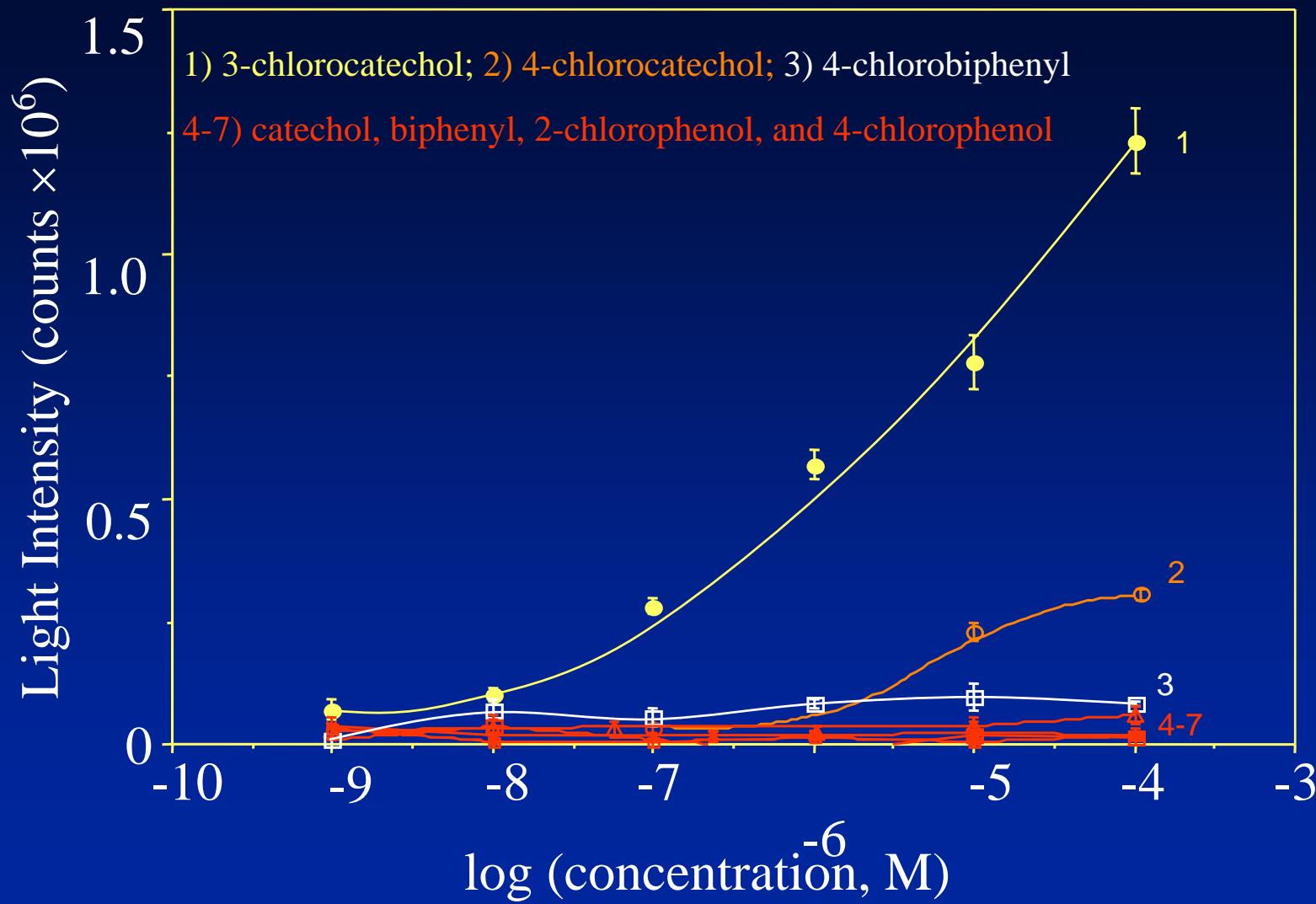
Clc Operon and 3-chlorocatechol Pathway



Calibration Curves



Selectivity Study



Challenges in Environmental Sample Analysis of Chlorocatechol

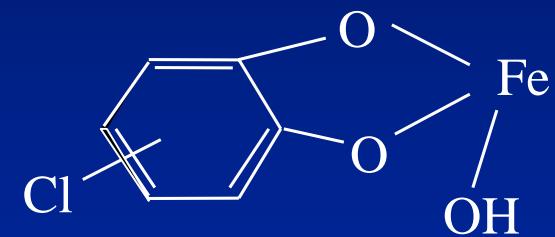
1. Extraction Method

- Free Chlorocatechol

Can be extracted by organic solvent

- Bound Chlorocatechol

Difficult to be extracted

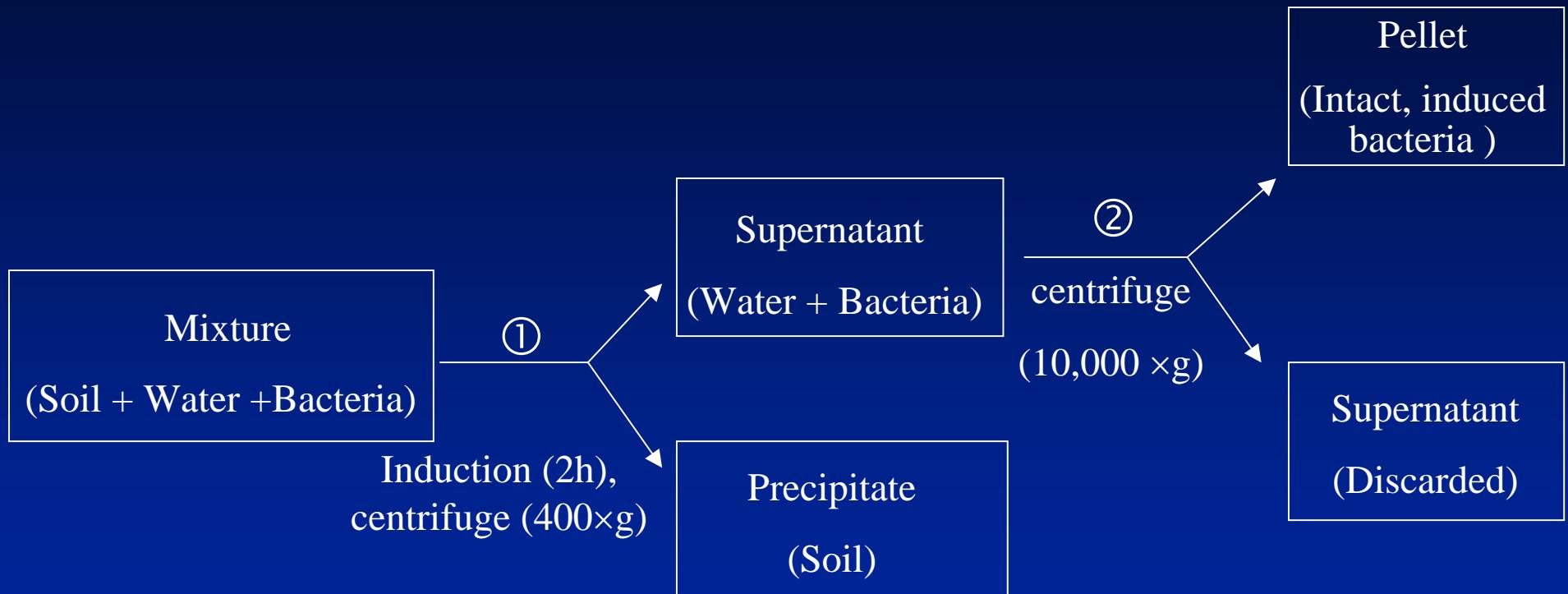


2. Matrix Effect

Selected Chemical Characteristics of Soils

Soils	Organic carbon (%)	Electrical conductivity ($\mu\text{mhos}\cdot\text{cm}^{-1}$)	pH	Oxalate extractable		
				P	Al	Fe
				(mg kg $^{-1}$)		
Acid washed sand	0.02	12	6.1	8	32	54
Maury silt loam	3.3	53	5.0	704	1894	4126
Woolper	7.5	38	5.9	3203	3086	3203
Organic humus	15.3	1099	6.4	415	3575	1106

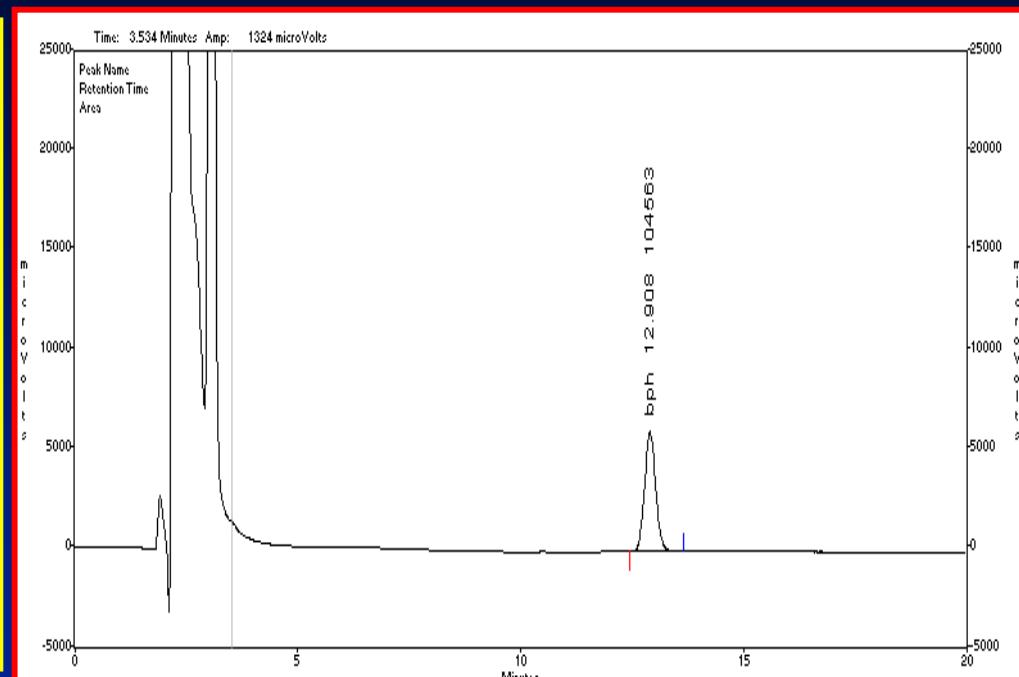
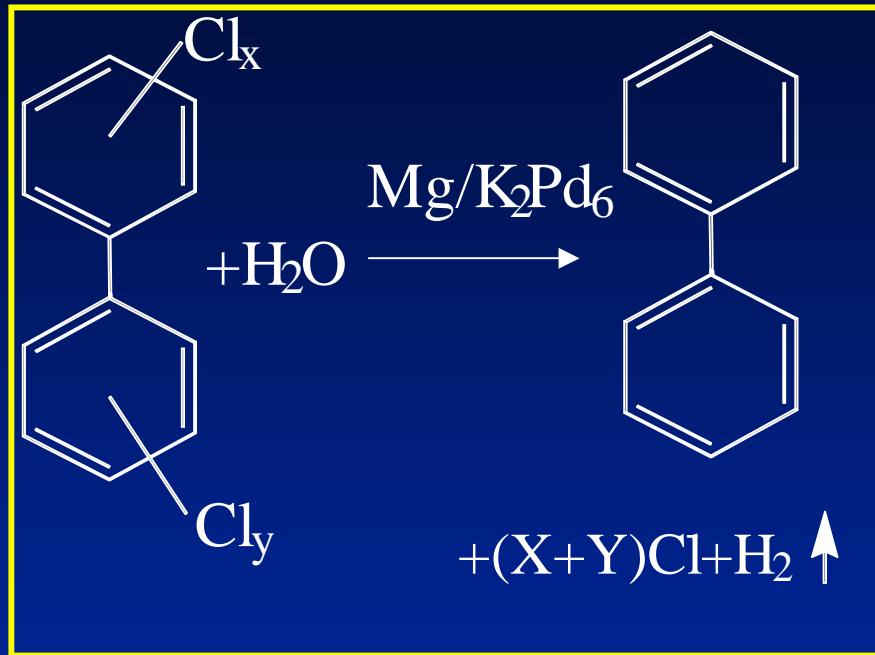
Optimized Protocol for Soil Analysis



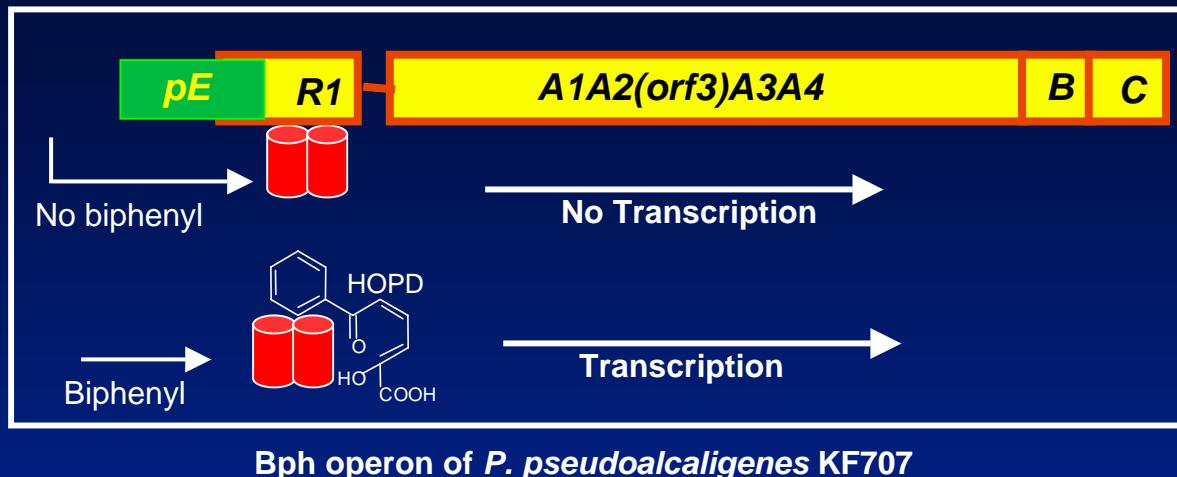
Results of Soils

Sample matrix	HPLC		Bacterial sensing system	
	Theoretical Value (mg•kg ⁻¹)	Experimental ± SD (mg•kg ⁻¹)	Theoretical Value (mg•kg ⁻¹)	Experimental ± SD (mg•kg ⁻¹)
Sand	0.5	0.49 ± 0.02	0.5	0.51 ± 0.02
	2.0	2.03 ± 0.04	2.0	1.95 ± 0.08
			10	9.75 ± 0.55
			50	52.0 ± 2.6
Woolper	0.5	0.0 ± 0.0	0.5	0.52 ± 0.04
	2.0	0.0 ± 0.0	2.0	1.90 ± 0.18
			10	9.55 ± 0.75
			50	54.0 ± 4.5
Maury	0.5	0.0 ± 0.0	0.5	0.43 ± 0.08
	2.0	0.0 ± 0.0	2.0	2.24 ± 0.30
Organic potting soil	0.5	0.0 ± 0.0	0.5	0.43 ± 0.09
	2.0	0.0 ± 0.0	2.0	1.73 ± 0.28

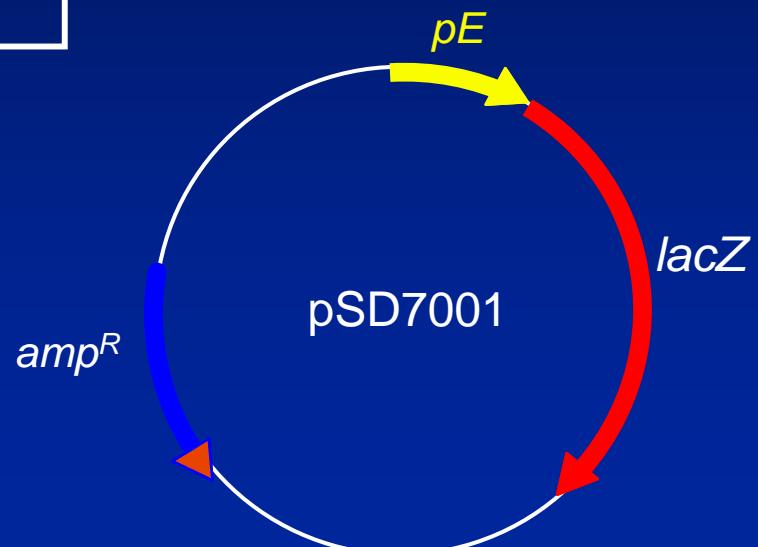
Detection of PCBs Based on Dechlorination Followed by whole cell Sensing



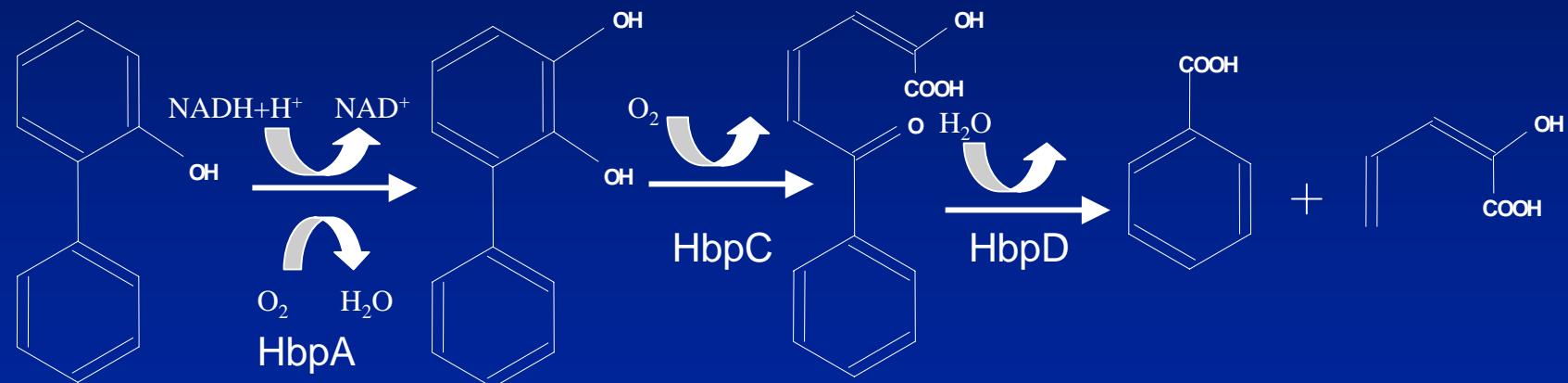
Whole cell Sensing of Biphenyl Based on *bph* Operon from *P. pseudoalcaligenes* KF707



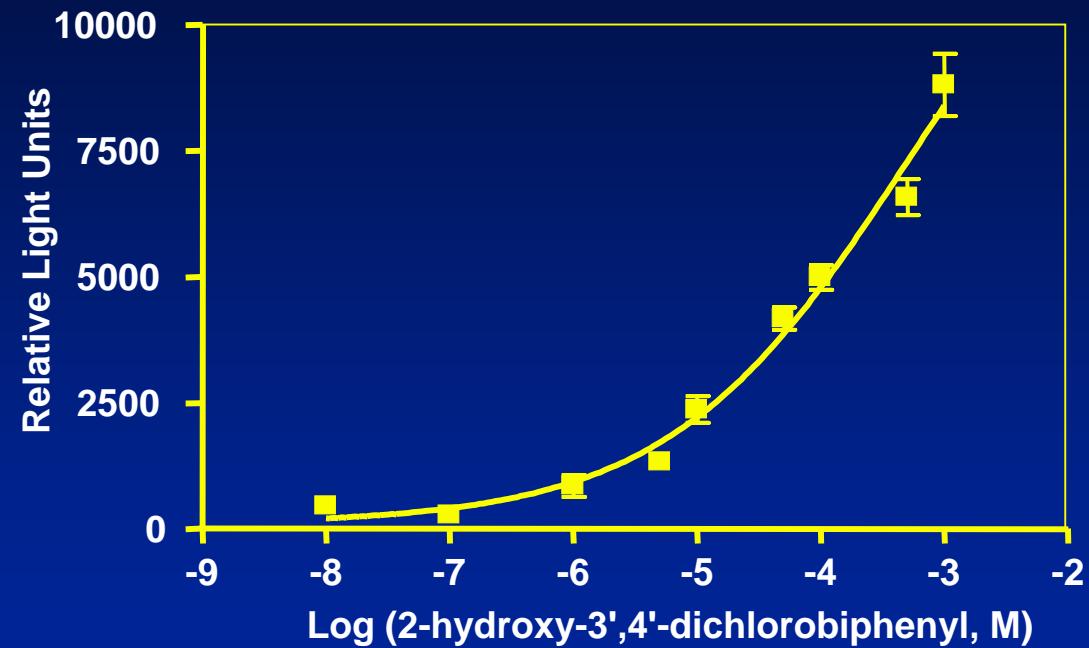
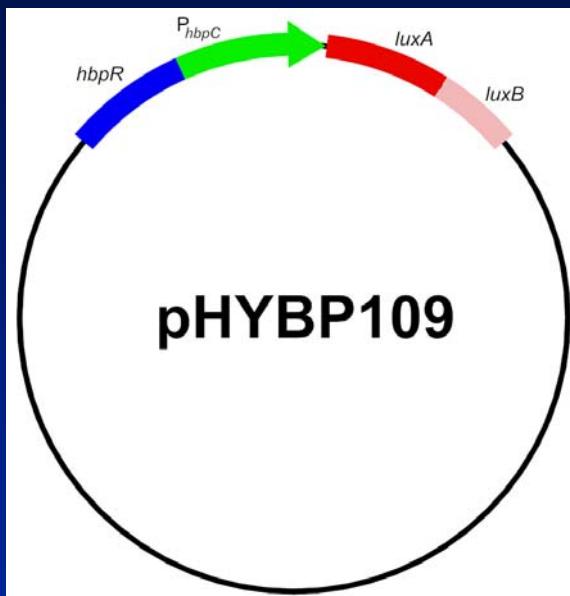
Biphenyl mol/L	Increase in Light Intensity (%)
1×10^{-5}	81.9 ± 9.5
1×10^{-6}	37 ± 6.2
1×10^{-7}	21 ± 4.8



Degradation Pathway of Hydroxylated-Biphenyl in the strain *Pseudomonas azelaica* HBP1



Whole Cell-Based Sensing System for Hydroxylated PCBs



Collaborators

Leonidas Bachas

Marc Madou

Barry Rosen

Jan Roelof van der Meer

Acknowledgments

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